



Isotope Analysis for Source Exploration and Degradation Assessment of Nitrate

Groundwater is often heavily polluted with nitrate of unknown origin. Based on isotope analyses ($^{15}\text{N}/^{14}\text{N}$ as $\delta^{15}\text{N}$, $^{18}\text{O}/^{16}\text{O}$ as $\delta^{18}\text{O}$) nitrate sources (e.g. atmospheric deposition, mineral fertilizers, organic fertilizers, sewage) can be identified. More precise source allocation can be provided by the additional isotope analysis of boron ($^{11}\text{B}/^{10}\text{B}$ as $\delta^{11}\text{B}$).

During denitrification, heavy nitrogen and oxygen isotopes (^{15}N , ^{18}O) accumulate in the residual nitrate, resulting in specific changes in $\delta^{15}\text{N}$ and $\delta^{18}\text{O}$ values. With isotope analyses of sulfate ($^{34}\text{S}/^{32}\text{S}$ as $\delta^{34}\text{S}$, $\delta^{18}\text{O}$) the degradation processes can be further evaluated. The ratio of nitrogen/argon (N_2/Ar) points to foregoing denitrification even in nitrate-free groundwater.



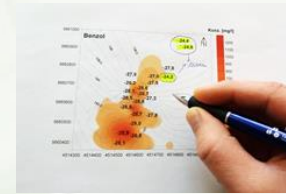
Sampling



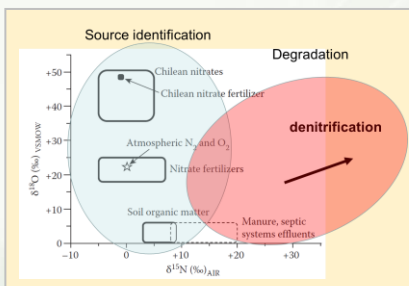
Preparation



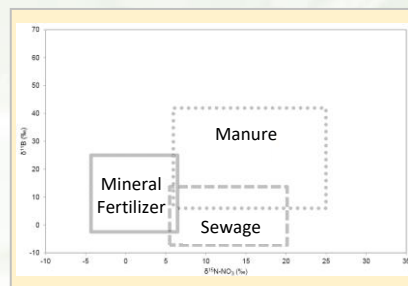
Analysis



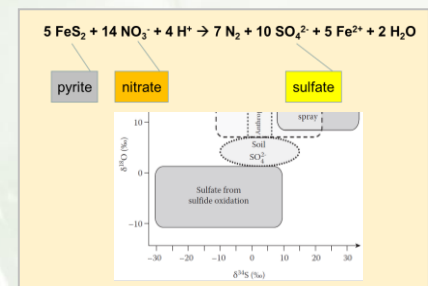
Expertise



Sources of nitrate have specific nitrogen and oxygen isotope values ($\delta^{15}\text{N}$, $\delta^{18}\text{O}$). Increasingly positive $\delta^{15}\text{N}$ and $\delta^{18}\text{O}$ values prove nitrate degradation.



Plotting isotope values of boron ($\delta^{11}\text{B}$) and nitrate-nitrogen ($\delta^{15}\text{N}$) allows for valid identification of nitrate sources.



Nitrate can be degraded by chemolithotrophic pyrite oxidation. The resulting sulfate exhibits particularly negative sulfur and oxygen isotope ratios ($\delta^{34}\text{S}$, $\delta^{18}\text{O}$).

Workflow

- We develop an investigation concept taking into account the site-specific framework conditions.
- We provide sampling flasks and sampling protocols.
- We perform isotope analysis of nitrate, boron and sulfate as well as the concentrations of N_2 & Ar.
- We interpret the data to evaluate nitrate sources and degradation processes.

Costs

- $^{15}\text{N}/^{14}\text{N}$, $^{18}\text{O}/^{16}\text{O}$: 175 – 240 € per sample
- $^{11}\text{B}/^{10}\text{B}$: 330 – 400 € per sample
- $^{34}\text{S}/^{32}\text{S}$, $^{18}\text{O}/^{16}\text{O}$: 175 – 240 € per sample
- N_2/Ar ratio: on request

Benefit

- Differentiation of nitrate sources
- Detection and quantification of nitrate degradation

Further reading

- Xu S, Kang P, Sun Y (2016): A stable isotope approach and its application for identifying nitrate source and transformation process in water. Environ. Sci. Pollut. Res. 23: 1133-1148.
- Widory D, Petelet-Giraud E, Négrel P, Ladouche B (2005) Tracking the source of nitrate in groundwater using coupled nitrogen and boron isotopes: a synthesis. Environ. Sci. Technol. 39: 539-548.
- Hosono T, Tokunaga T, Tsushima A, Shimada J (2014) Combined use of $\delta^{13}\text{C}$, $\delta^{15}\text{N}$, and $\delta^{34}\text{S}$ tracers to study anaerobic bacterial processes in groundwater flow systems. Water Res. 54: 284-296.