



## Isotope Analysis for Source Exploration and Degradation Assessment of Nitrate

Groundwater is often heavily polluted with nitrate of unknown origin. Based on isotope analyses ( $^{15}\text{N}/^{14}\text{N}$  as  $\delta^{15}\text{N}$ ,  $^{18}\text{O}/^{16}\text{O}$  as  $\delta^{18}\text{O}$ ) nitrate sources (e.g. atmospheric deposition, mineral fertilizers, organic fertilizers, sewage) can be identified. More precise source allocation can be provided by the additional isotope analysis of boron ( $^{11}\text{B}/^{10}\text{B}$  as  $\delta^{11}\text{B}$ ).

During denitrification, heavy nitrogen and oxygen isotopes ( $^{15}\text{N}$ ,  $^{18}\text{O}$ ) accumulate in the residual nitrate, resulting in specific changes in  $\delta^{15}\text{N}$  and  $\delta^{18}\text{O}$  values. With isotope analyses of sulfate ( $^{34}\text{S}/^{32}\text{S}$  as  $\delta^{34}\text{S}$ ,  $\delta^{18}\text{O}$ ) the degradation processes can be further evaluated. The ratio of nitrogen/argon ( $\text{N}_2/\text{Ar}$ ) points to foregoing denitrification even in nitrate-free groundwater.



**Sampling**



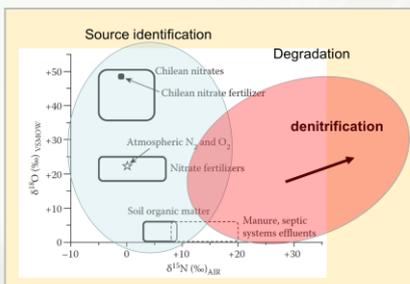
**Preparation**



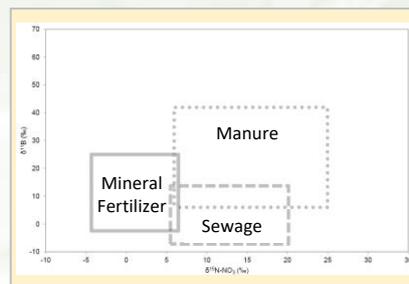
**Analysis**



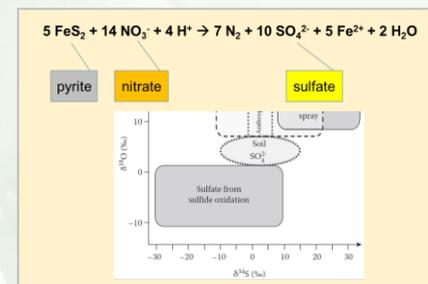
**Expertise**



**Sources of nitrate have specific nitrogen and oxygen isotope values ( $\delta^{15}\text{N}$ ,  $\delta^{18}\text{O}$ ). Increasingly positive  $\delta^{15}\text{N}$  and  $\delta^{18}\text{O}$  values prove nitrate degradation.**



**Plotting isotope values of boron ( $\delta^{11}\text{B}$ ) and nitrate-nitrogen ( $\delta^{15}\text{N}$ ) allows for valid identification of nitrate sources.**



**Nitrate can be degraded by chemolithotrophic pyrite oxidation. The resulting sulfate exhibits particularly negative sulfur and oxygen isotope ratios ( $\delta^{34}\text{S}$ ,  $\delta^{18}\text{O}$ ).**

### Workflow

- We develop an investigation concept taking into account the site-specific framework conditions.
- We provide sampling flasks and sampling protocols.
- We perform isotope analysis of nitrate, boron and sulfate as well as the concentrations of  $\text{N}_2$  & Ar.
- We interpret the data to evaluate nitrate sources and degradation processes.

### Costs

- $^{15}\text{N}/^{14}\text{N}$ ,  $^{18}\text{O}/^{16}\text{O}$ : 175 – 240 € per sample
- $^{11}\text{B}/^{10}\text{B}$ : 330 – 400 € per sample
- $^{34}\text{S}/^{32}\text{S}$ ,  $^{18}\text{O}/^{16}\text{O}$ : 175 – 240 € per sample
- $\text{N}_2/\text{Ar}$  ratio: on request

### Benefit

- Differentiation of nitrate sources
- Detection and quantification of nitrate degradation

### Further reading

- Xu S, Kang P, Sun Y (2016): A stable isotope approach and its application for identifying nitrate source and transformation process in water. Environ. Sci. Pollut. Res. 23: 1133-1148.
- Widory D, Petelet-Giraud E, Négrel P, Ladouche B (2005) Tracking the source of nitrate in groundwater using coupled nitrogen and boron isotopes: a synthesis. Environ. Sci. Technol. 39: 539-548.
- Hosono T, Tokunaga T, Tsushima A, Shimada J (2014) Combined use of  $\delta^{13}\text{C}$ ,  $\delta^{15}\text{N}$ , and  $\delta^{34}\text{S}$  tracers to study anaerobic bacterial processes in groundwater flow systems. Water Res. 54: 284-296.